A Facile Construction of Withanolide Side Chains: Synthesis of Minabeolide-3

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A stereocontrolled construction of withanolide side chains, withanolide R-type 17, and 27-deoxywithaferin A-type 21, was accomplished from 20-carboxaldehyde 6. The key features of the construction are based on the stereoselective synthesis of the furylcarbinol 12 and its transformation into pyranone derivatives. Synthesis of minabeolide-3 28 was achieved from the key intermediate 18.

Withanolides, a group of highly oxygenated ergostane type steroids possessing a δ -lactone in the side chain, have been isolated from the plants of the Solanaceae family1 and also marine organisms.² Due to their novel structures and interesting biological activities,3 such as antitumor and insect antifeedant properties, withanolides have been recognized as attractive synthetic targets.^{1,4} As part of our continuing studies on the synthesis of physiologically active steroidal side chains using furan and butenolide as synthons,⁵ we herein report a new method for the construction of the withanolide side chain bearing a (22R)δ-lactone moiety and its application to the synthesis of minabeolide-3, which was isolated from a soft coral, Minabea sp., recently.² Although several syntheses of withanolides have been published to date,4 many steps have been required to construct an unsaturated δ -lactone moiety in the synthesis of the withaferin A-type side chain. 4a,b Our synthetic strategy of withanolide side chains is based on the stereoselective synthesis of the (22S)-furylcarbinol 2, which could be easily transformed into the 27-deoxywith a ferin A type 5 via the pyranone 3 and the with a nolide R type 4 (Figure 1).

Results and Discussion

We began to explore the stereocontrolled synthesis of the (22S)-furylcarbinol 11 as a model compound (Scheme I). Our previous result^{5e} indicated that nucleophilic addition of 2-lithiofuran 7 to the 20-carboxaldehyde 6 could provide mainly the (22R)-isomer 9, a Cram product. For the synthesis of withanolide side chains, it was necessary to prepare the anti-Cram product 11, and this was at-

Table I. Reduction of 13 with Reducing Agents

reducing agent	ratio of products 9:11	% yield
DIBAL	1:1.8	91.2
$Zn(BH_4)_2$	1:2.2	86.9
L-Selectride	1:3.2	88.8
LiAlH ₄	1:3.6	91.1
NaBH.	1:3.7	90.0
Red-Al	1:3.8	93.1

tempted using Yamamoto's method.6 Unfortunately. addition of 2-lithiofuran⁷ to the aldehyde 6⁸ in the presence of methylaluminum bis(2,4,6-tri-tert-butylphenoxide)⁶ proceeded with the same selectivity as the previous result to give the isomers 9 and 11 in a ratio of 2:1, respectively. Therefore, conversion of 9 to 119 was examined by oxidation/reduction sequence via the acyl furan 13. Alcohol 9 was oxidized with pyridinium chlorochromate (PCC)^{5e} to give the ketone 13. Several reducing agents were surveyed for the reduction of 13, and the results are shown in Table I. Interestingly, the reductions proceeded with moderate Cram selectivity to afford the isomer 11 predominantly in a ratio of 1.8-3.8:1, although reduction of 22-oxosteroids having saturated side chains provides mainly anti-Cram products. 10 This selectivity agrees with the previous observation in the reduction of 22-oxo- Δ^{23} steroids with L-Selectride.11

On the basis of these results, we prepared the key intermediate 12 for the construction of the withanolide side chains as follows. Addition of 2-lithio-3,4-dimethylfuran 8, prepared by lithiation of 3,4-dimethylfuran, 12 to the aldehyde 6 produced the furylcarbinols 10 and 12 in a ratio of 3.2:1 (79.5%). Oxidation of the alcohol 10 with PCC employed conditions similar to the oxidation of 9; however, it gave a mixture of products. Several oxidation conditions, such as dimethyl sulfoxide (DMSO)-oxalyl chloride¹³ and manganese dioxide,14 were investigated, and we found that 10 was successfully oxidized with use of 2,3-dichloro-5,6dicyano-1,4-benzoquinone (DDQ)¹⁵ in dioxane to afford the desired acyl furan 14 in 61.4% yield. Reduction of the ketone 14 with lithium aluminum hydride in tetrahydrofuran (THF) proceeded with high diastereoselectivity to

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Scheme I

17 R¹=OH, R²=H 18 R1=OAc, R2=H

5 R1+R2=H2

Figure 1.

furnish the desired (22S)-furylcarbinol 12 together with 10 in a ratio of 16:1.

With the requisite furylcarbinol 12 in hand, we embarked on the synthesis of the withanolide side chains as shown in Scheme II. Treatment of 12 with N-bromosuccinimide (NBS)¹⁶ in aqueous THF brought about the

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Figure 2.

ring enlargement to give the lactol 15 in 84.4% yield. Further oxidation of 15 with PCC provided the relatively unstable δ-lactone 16 which, without purification, was reduced with sodium borohydride to afford the alcohol 17, the withanolide R-type side chain, as a sole product in 73.4% yield from 15. The ¹H NMR spectrum of its acetate 18, prepared by acetylation of the alcohol 17, showed good accordance with that of with anolide R monoacetate. 17 The observed selectivity can be explained by assuming that the reduction of the ketone 16 would occur from the opposite side of steroid nucleus (Figure 2).

Reductive deacetoxylation of 18 with zinc-amalgam¹⁸ in ethereal hydrogen chloride gave the olefin 19 in 20.8%

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Scheme III

yield, together with the starting material (76.7%). Side reactions were observed under forced reaction conditions in which the cyclopropyl moiety of 18 was opened to produce the chloride 20 as a major product. Isomerization of the olefin 19 with 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU)^{5b} furnished the α,β -unsaturated lactone 21,¹⁹ the 27-deoxywithaferin A side chain, whose spectroscopic data were identical with those reported.4c

Establishing a new method for the construction of the with anolide side chains, we next focused our attention on the synthesis of minabeolide-3 28 from the lactone 18 (Scheme III). Due to the difficulty observed in the reductive deacetoxylation of 18, ring opening of the cyclopropane in 18 was carried out prior to further transformation of the side-chain moiety. Acid treatment²⁰ of 18 with p-toluenesulfonic acid in aqueous dioxane afforded the olefinic alcohol 22 whose protection with tert-butylchlorodiphenylsilane21 gave the silyl ether 23 quantitatively. Allyl acetate 23 was subjected to the reductive deacetoxylation to give the olefin 24 in 80% yield. Upon treatment with hydrogen fluoride. 22 24 provided the alcohol 25 which was isomerized with DBU to the α,β -unsaturated lactone 26. Finally, Swern oxidation of 26 gave the ketone 27 which was treated with oxalic acid in ethanol to afford minabeolide-3 28 in 88.4% yield. The spectral data of the synthetic minabeolide-3 showed good agreement with those reported.2

Thus, we have developed a synthesis of the withanolide side chains by a facile method that could be applied to the synthesis of highly oxygenated withanolides.

Experimental Section

Melting points were measured with a Yanagimoto MP apparatus and are uncorrected. IR spectra were recorded on a Hitachi 260-10 spectrophotometer. ¹H NMR spectra were obtained for solutions in CDCl₃ on a JEOL GSX-270 instrument, and chemical shifts are reported on the δ -scale from internal TMS. Mass spectra were measured with a JEOL JMS D-300 spectrometer. Optical rotations were taken with a JASCO DIP-360 polarimeter.

Addition of 2-Lithiofuran to the 20-Carboxaldehyde 6 with MAT. To a stirred solution of methylaluminum bis(2,4,6-tritert-butylphenoxide), prepared from 2,4,6-tri-tert-butylphenol (944 mg, 3.49 mmol) and Me₃Al (1.76 mL of a 0.99 M hexane solution, 1.74 mmol) in CH₂Cl₂ (12 mL) according to Yamamoto's method, was added a solution of the aldehyde 68 (200 mg, 0.58) mmol) in THF (3 mL) followed by 2-lithiofuran, prepared from furan (0.17 mL, 2.32 mmol) and n-BuLi (1.13 mL of a 1.65 M hexane solution, 1.86 mmol) in THF (0.5 mL), at -78 °C. The reaction mixture was stirred for 2 h at room temperature, and aqueous saturated NH₄Cl solution was added. The product was extracted with CH2Cl2, and the organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (16:1, v/v) as eluent to afford the (22R)-alcohol 9 (110 mg, 45.5%) as colorless needles: mp 134-136 °C (hexane–AcOEt); IR $\nu_{\rm max}$ 3360 cm⁻¹; ¹H NMR δ 0.77 (s, 3 H, $18-CH_3$), 0.88 (d, J = 6.7 Hz, 3 H, $21-CH_3$), 1.03 (s, 3 H, $19-CH_3$), $2.78 (t, J = 3.0 \text{ Hz}, 1 \text{ H}, 6\text{-CH}), 3.33 (s, 3 \text{ H}, OCH_3), 4.86 (br s, 3 \text{ H}, OCH_3)$ 1 H, 22-CH), 6.21 (d, J = 3.1 Hz, 1 H, 24-CH), 6.33 (dd, J = 1.8and 3.1 Hz, 1 H, 25-CH), 7.35 (d, J = 1.8 Hz, 1 H, 26-CH); HRMS calcd for $C_{27}H_{40}O_3$ (M⁺) 412.2993, found (M⁺) 412.2997. Anal. Calcd for C₂₇H₄₀O₃: C, 78.59; H, 9.77. Found C, 78.59; H, 9.92. The further elution gave the (22S)-alcohol 11 (53 mg, 21.9%) as colorless needles: mp 139–141 °C (hexane–AcOEt); IR $\nu_{\rm max}$ 3350 cm⁻¹; 1 H NMR δ 0.76 (s, 3 H, 18-CH₃), 1.01 (s, 3 H, 19-CH₃), 1.02 $(d, J = 6.1 \text{ Hz}, 3 \text{ H}, 21\text{-CH}_3), 2.76 (t, J = 2.7 \text{ Hz}, 1 \text{ H}, 6\text{-CH}), 3.32$ (s, 3 H, OCH₃), 4.81 (d, J = 3.7 Hz, 1 H, 22-CH), 6.21 (d, J = 3.1Hz, 1 H, 24-CH), 6.33 (dd, J = 1.8 and 3.1 Hz, 1 H, 25-CH), 7.35 (d, J = 1.8 Hz, 1 H, 26-CH); HRMS calcd for $C_{27}H_{40}O_3$ (M⁺) 412.2993, found (M⁺) 412.2975. Anal. Calcd for C₂₇H₄₀O₃: C, 78.59; H, 9.77. Found C, 78.46; H, 9.91.

(20S,23Z,25Z)-23,26-Epoxy-6 β -methoxy-3 α ,5-cyclo-27nor-5α-cholesta-23,25-dien-22-one (13). To a stirred suspension of PCC (56 mg, 0.26 mmol), Celite (60 mg), and anhydrous sodium acetate (22 mg, 0.26 mmol) in CH₂Cl₂ (0.2 mL) was added a solution of 9 in CH₂Cl₂ (0.4 mL) at room temperature. Stirring was continued for 1 h, and the reaction mixture was diluted with Et₂O (5 mL). The organic layer was decanted and evaporated to give a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (20:1, v/v) as eluent to afford the ketone 13 (25.3 mg, 70.9%) as a colorless gum: IR $\nu_{\rm max}$ 1660 cm⁻¹; ¹H NMR δ 0.83 (s, 3 H, 18-CH₃), 1.04 (s, 3 H, 19-CH₃), 1.22 $(d, J = 6.7 \text{ Hz}, 3 \text{ H}, 21\text{-CH}_3), 2.77 \text{ (t, } J = 2.7 \text{ Hz}, 1 \text{ H}, 6\text{-CH}), 3.26$ $(dq, J = 6.7 \text{ and } 10.4 \text{ Hz}, 1 \text{ H}, 20\text{-CH}), 3.32 (s, 3 \text{ H}, OCH_3), 6.53$ (dd, J = 1.8 and 3.7 Hz, 1 H, 25-CH), 7.20 (d, J = 3.7 Hz, 1 H, 1 H)26-CH), 7.60 (d, J = 1.8 Hz, 1 H, 24-CH); HRMS calcd for C_{27} $H_{38}O_3$ (M⁺) 410.2821, found (M⁺) 410.2822.

Reduction of the Ketone 13. General Procedure. To a solution of 13 (50 mg, 0.12 mmol) in appropriate solvent was added reducing agent at -78 °C under argon atmosphere. The reaction was warmed to room temperature, and general workup was performed. Purification of the crude product by column chromatography on silica gel using hexane-AcOEt (10:1, v/v) as eluent afforded a mixture of the isomeric alcohols 9 and 11. Ratio of the products was estimated by the ¹H NMR spectrum of the

DIBAL Reduction of 13. Reduction of 13 with dissobutylaluminum hydride (DIBAL) (0.24 mL of a 1 M toluene solution, 0.24 mmol) in toluene (5 mL) was performed as above to afford a mixture of 9 and 11 (45.8 mg, 91.2%) in a ratio of 1:1.8.

 $\mathbf{Zn}(\mathbf{BH_4})_2$ Reduction of 13. Reduction of 13 with zinc borohydride [Zn(BH₄)₂] (2.73 mL of a 0.14M Et₂O solution, 0.38 mmol) in Et₂O (5 mL) was performed as above to afford a mixture of 9 and 11 (43.6 mg, 86.9%) in a ratio of 1:2.2.

L-Selectride Reduction of 13. Reduction of 13 with L-Selectide (0.24 mL of a 1 M THF solution, 0.24 mmol) in THF (5 mL) was performed as above to afford a mixture of 9 and 11 (44.6 mg, 88.8%) in a ratio of 1:3.2.

LiAlH, Reduction of 13. Reduction of 13 with lithium aluminum hydride (LiAlH₄) (7 mg, 0.18 mmol) in THF (2 mL) was performed as above to afford a mixture of 9 and 11 (45.8 mg, 91.1%) in a ratio of 1:3.6.

NaBH, Reduction of 13. Reduction of 13 with sodium borohydride (NaBH₄) (7 mg, 0.18 mmol) in MeOH (5 mL) was performed as above to afford a mixture of 9 and 11 (45.2 mg, 90.0%) in a ratio of 1:3.7.

Red-Al Reduction of 13. Reduction of 13 with Red-Al (0.04) mL of a 3.4 M toluene solution, 0.12 mmol) in toluene (2 mL) was

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performed as above to afford a mixture of 9 and 11 (46.8 mg. 93.1%) in a ratio of 1:3.8.

Addition of 2-Lithio-3,4-dimethylfuran to the 20-Carboxaldehyde 6. To a stirred solution of 2-lithio-3,4-dimethylfuran, prepared from 3,4-dimethylfuran¹² (3.15 g, 32.8 mmol) and n-BuLi (18 mL of a 1.64 M hexane solution, 29.5 mmol) in THF (40 mL), was added a solution of the aldehyde 6 (5.63 g, 16.4 mmol) in THF (40 mL) at -78 °C. The reaction mixture was warmed to room temperature, and aqueous saturated NH₄Cl solution was added. Concentration of the solvent afforded an oil, which was extracted with AcOEt. The organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (16:1, v/v) as eluent to afford the (22R)-alcohol 10 (4.38 g, 60.7%) as a colorless gum: $[\alpha]^{24}_{\rm D}$ +32.2° (c 1.20, CHCl₃); IR $\nu_{\rm max}$ 3370 cm⁻¹; ¹H NMR δ 0.73 (s, 3 H, 18-CH₃), 0.98 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.02 (s, 3 H, 19-CH₃), 1.92 and 1.93 (2 \times s, 2×3 H, $CH_3C=CCH_3$), 2.78 (t, J = 3.1 Hz, 1 H, 6-CH), 3.33 (s, 3 H, OCH₃), 4.83 (d, J = 2.4 Hz, 1 H, 22-CH), 7.08 (br s, 1 H, C=CH); HRMS calcd for C₂₉H₄₄O₃ (M⁺) 440.3289, found (M⁺) 440.3287. The further elution gave the (22S)-alcohol 12 (1.36 g, 18.8%) as colorless needles: $[\alpha]^{24}_{D}$ +43.6° (c 1.05, CHCl₃); mp 128–129 °C (hexane–AcOEt); IR $\nu_{\rm max}$ 3450 cm⁻¹; ¹H NMR δ 0.77 (s, 3 H, 18-CH₃), 1.01 (s, 3 H, 19-CH₃), 1.13 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.92 (d, J = 1.8 Hz, 3 H, C=CCH₃), 1.96 (s, 3 H, C= CCH_3), 2.76 (t, J = 3.1 Hz, 1 H, 6-CH), 3.32 (s, 3 H, OCH₃), 4.72 (d, J = 4.3 Hz, 1 H, 22-CH), 7.09 (d, J = 1.8 Hz, 1 H, C=CH);HRMS calcd for C₂₉H₄₄O₃ (M⁺) 440.3289, found (M⁺) 440.3289.

Oxidation of the Alcohol 10. To a stirred suspension of DDQ (12.9 mg, 0.05 mmol) in dioxane (0.3 mL) was added a solution of 10 (20 mg, 0.045 mmol) in dioxane (0.5 mL) under argon at room temperature. Stirring was continued for 30 min, and the solvent was concentrated to give a residue. The residue was diluted with CH2Cl2, and the precipitate was filtered off. Evaporation of the filtrate afforded a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (20:1, v/v) as eluent to afford the ketone 14 (12.2 mg. 61.4%) as colorless plates: $[\alpha]^{21}_{\rm D}$ +57.6° (c 1.47, CHCl₃); mp 161–161.5 °C (hexane–AcOEt); IR $\nu_{\rm max}$ 1660 cm⁻¹; ¹H NMR δ 0.82 (s, 3 H, 18-CH₃), 1.03 (s, 3 H, 19-CH₃), 1.17 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.97 (d, J = 1.2 Hz, 3 H, C=CCH₃), 2.30 (s, 3 H, C=CCH₃), 2.77 $(t, J = 3.1 \text{ Hz}, 1 \text{ H}, 6\text{-CH}), 3.32 \text{ (s, 3 H, OCH}_3), 3.41 \text{ (dq, } J = 6.7)$ and 10.4 Hz, 1 H, 20-CH), 7.22 (br s, 1 H, C=CH); HRMS calcd for C₂₉H₄₂O₃ (M⁺) 438.3132, found (M⁺) 438.3130. Anal. Calcd for C₂₉H₄₂O₃: C, 79.40; H, 9.65. Found: C, 79.15; H, 9.83.

Reduction of the Ketone 14. To a stirred suspension of lithium aluminum hydride (768 mg, 20 mmol) in THF (20 mL) was added a solution of 14 (1.77 g, 4 mmol) in THF (40 mL) at -78 °C under argon atmosphere. The reaction was warmed to room temperature, and aqueous 1 M sodium hydroxide solution was added slowly. The precipitate was filtered off, and the filtrate was concentrated to give a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (10:1, v/v) as eluent, affording the isomeric alcohols 10 (104 mg, 5.8%) and 12 (1.67 g, 93.8%).

NBS Oxidation of the Alcohol 12. To a solution of 12 (1.15 g, 2.61 mmol) and anhydrous sodium acetate (235 mg, 2.87 mmol) in aqueous THF (15 mL, THF:H₂O = 4:1) was added portionwise N-bromosuccinimide (512 mg, 2.87 mmol) at 0 °C, and stirring was continued for 30 min. After addition of aqueous 10% KI solution and then aqueous saturated sodium thiosulfate solution to the reaction mixture, the product was extracted with AcOEt. The organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent afforded a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (20:1, v/v) as eluent to afford the lactol 15 (1.0 g, 84.4%) as a colorless gum: IR $\nu_{\rm max}$ 3380, 1670 cm⁻¹; ¹H NMR δ 0.73 (s, 3 H, $18-CH_3$, 1.02 (s, 3 H, $19-CH_3$), 1.02 (d, J = 6.7 Hz, 3 H, $21-CH_3$), 1.77 and 1.96 (2 × br s, 2 × 3 H, CH₃C—CCH₃), 2.77 (t, J = 2.4Hz, 1 H, 6-CH), 3.20 (br s, 1 H, OH), 3.32 (s, 3 H, OCH₃), 3.9-4.45 (m, 1 H, 22-CH), 5.44 (br s, 1 H, CHOH); HRMS calcd for C₂₉- $H_{44}O_4$ (M⁺) 456.3240, found (M⁺) 456.3245. Anal. Calcd for C₂₉H₄₄O₄: C, 76.27; H, 9.71. Found C, 76.52; H, 9.74.

(20S, 22S, 23S)-23-Hydroxy-6eta-methoxy-3lpha,5-cyclo-5lphawith-24-enolide (17). To a stirred suspension of PCC (1.39 g, 6.45 mmol), Celite (1 g), and anhydrous sodium acetate (530 mg,

6.45 mmol) in CH₂Cl₂ (5 mL) was added a solution of 15 (980 mg, 2.15 mmol) in CH₂Cl₂ (7 mL) at room temperature. Stirring was continued for 2 h, and the reaction mixture was diluted with Et₂O (60 mL). The organic layer was decanted and evaporated to give a residue. A small amount of the crude product was purified by column chromatography on silica gel using hexane-AcOEt (20:1, v/v) as eluent to afford the lactone 16 as a colorless gum: IR $\nu_{\rm max}$ 1715, 1680 cm⁻¹; ¹H NMR δ 0.71 (s, 3 H, 18-CH₃), 0.96 (d, J =6.7 Hz, 3 H, 21-CH₃), 0.99 (s, 3 H, 19-CH₃), 1.96 and 2.15 (2 \times br s, 2×3 H, 27-CH₃ and 28-CH₃), 2.75 (t, J = 2.4 Hz, 1 H, 6-CH), $3.30 (s, 3 H, OCH_3), 4.85 (d, J = 2.5 Hz, 1 H, 22-CH);$ HRMS calcd for $C_{29}H_{42}O_4$ (M⁺) 454.3082, found (M⁺) 454.3077. The crude product obtained above was used without further purification due to the instability of 16. To a solution of 16 in MeOH (8 mL) was added portionwise sodium borohydride (205 mg, 5.38 mmol) at 0 °C, and the reaction mixture was stirred for 30 min. Aqueous saturated NH₄Cl solution was added slowly, and the solvent was concentrated. The product was extracted with AcOEt, and the organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (4:1, v/v) as eluent to afford the title alcohol 17 (723 mg, 73.4% from 15) as a colorless gum: $[\alpha]^{22}_{D}$ +71.2° (c 0.13, CHCl₃); IR ν_{max} 3400, 1700 cm⁻¹; ¹H NMR δ 0.78 (s, 3 H, 18-CH₃), 1.02 (s, 3 H, 19-CH₃), 1.21 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.90 and 2.04 (2 × br s, 2 × 3 H, 27-CH₃ and 28-CH₃), 2.78 (t, J = 2.4 Hz, 1 H, 6-CH), 3.33 (s, 3 H, OCH₃), 3.97 (br s, 1 H, 23-CH), 4.20 (dd, J = 1.8 and 6.1 Hz, 1 H, 22-CH); HRMS calcd for C₂₂H₄₄O₄ (M⁺) 456.3240, found (M^+) 456.3243. Anal. Calcd for $C_{29}H_{44}O_4$: C, 76.27; H, 9.71. Found: C, 75.78; H, 9.71.

(20S,22S,23S)-23-Acetoxy-6 β -methoxy-3 α ,5-cyclo-5 α with-24-enolide (18). A mixture of the alcohol 17 (120 mg, 0.26 mmol), acetic anhydride (75 µL, 0.79 mmol), 4-(dimethylamino)pyridine (3.2 mg, 0.03 mmol), and pyridine (85 μ L, 1.05 mmol) in CH₂Cl₂ (2 mL) was stirred at 0 °C for 1 h. The reaction mixture was poured into water, and the product was extracted with CH₂Cl₂. The organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (7:1, v/v) as eluent to afford the acetate 18 (125 mg, 95.4%) as colorless plates: $[\alpha]^{23}_{D}$ +179.5° (c 2.72, CHCl₃); mp 194-195 °C (hexane-AcOEt); IR ν_{max} 1730, 1715 cm⁻¹; ¹H NMR δ 0.75 (s, 3 H, 18-CH₃), 1.02 (s, 3 H, 19-CH₃), 1.06 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.94 and 1.95 (2 × br s, 2 × 3 H, 27-CH₃ and 28-CH₃), 2.09 (s, 3 H, OAc), 2.77 (t, J = 2.4 Hz, 1 H, 6-CH), 3.32 (s, 3 H, OCH_3), 4.36 (dd, J = 1.8 and 5.5 Hz, 1 H, 22-CH), 5.47 (d, J = 1.8 Hz, 1 H, 23-CH); HRMS calcd for $C_{31}H_{48}O_5$ (M⁺) 498.3344, found (M⁺) 498.3339. Anal. Calcd for $C_{31}H_{46}O_5$: C, 74.66; H, 9.30. Found: C, 74.84; H, 9.50. The ¹H NMR spectrum of 18 in pyridine-d5 showed good accordance with that of withanolide R monoacetate: ¹⁵ ¹H NMR (pyridine- d_5) δ 0.66 (s, 3 H, $18-CH_3$, 1.15 (s, 3 H, 19-CH₃), 1.19 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.92 and 1.96 (2 \times br s, 2 \times 3 H, 27-CH₃ and 28-CH₃), 2.16 (s, 3 H, OAc), 2.74 (br s, 1 H, 6-CH), 3.31 (s, 3 H, OCH₃), 4.56 (dd, J = 1.8 and 5.5 Hz, 1 H, 22-CH), 5.77 (br s, 1 H, 23-CH).

 $(20S,22R,25\xi)$ -6 β -Methoxy-3 α ,5-cyclo-5 α -with-23-enolide (19). To a suspension of zinc amalgam (79 mg, 1.2 mmol) and 18 (30 mg, 0.06 mmol) in Et₂O (2 mL) was added hydrogen chloride (47 µL of a 2.5 M Et₂O solution, 0.12 mmol) over 15 min at -15 °C under argon atmosphere, and the reaction was warmed to room temperature. The inorganic material was filtered off, and the organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (7:1, v/v) as eluent to afford the olefin 19 (5.5 mg, 20.8%) as a colorless gum: IR $\nu_{\rm max}$ 1725 cm⁻¹; ¹H NMR δ 0.76 (s, 3 H, 18-CH₃), 0.86 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.03 (s, 3 H, 19-CH₃), 1.41 (d, J = 7.3 Hz, 3 H, 27-CH₃), 1.59 (s, 3 H, 28-CH₃), 2.78 (t, J = 2.4 Hz, 1 H, 6-CH), 2.85-3.0 (m, 1 H, 25-CH), 3.33 (s, 3 H, 25-CH)OCH₃), 4.91 (br s, 1 H, 22-CH), 5.50 (br s, 1 H, 23-CH); HRMS calcd for $C_{29}H_{44}O_3$ (M⁺) 440.3291, found (M⁺) 440.3292. The further elution gave starting material 18 (23 mg, 76.7%).

 $(20S,22R,25\xi)$ -3 β -Chlorowitha-5,23-dienolide (20). Reaction of 18 with hydrogen chloride (3.5 M Et₂O solution) was performed as above to afford the chloride 20 (95%) as a colorless gum: IR ν_{max} 1720 cm⁻¹; ¹H NMR δ 0.72 (s, 3 H, 18-CH₃), 0.86 (d, J = 6.7

Hz, 3 H, 21-CH₃), 1.03 (s, 3 H, 19-CH₃), 1.41 (d, J = 7.3 Hz, 3 H, 27-CH₃), 1.79 (s, 3 H, 28-CH₃), 2.85-3.0 (m, 2 H, 7-CH₂), 3.7-3.85 (m, 1 H, 3-CH), 4.91 (br s, 1 H, 22-CH), 5.37 (d, J = 5.5 Hz, 1 H, 6-CH), 5.51 (br s, 1 H, 23-CH); HRMS calcd for C₂₈-H₄₁O₂Cl (M⁺) 444.2793, found (M⁺) 444.2787. Anal. Calcd for C₂₈H₄₁O₂Cl: C, 75.56; H, 9.29. Found: C, 75.97; H, 9.65.

(20S,22R)- 6β -Methoxy- 3α ,5-cyclo- 5α -with-24-enolide (21). A mixture of the olefin 19 (5.5 mg, 0.01 mmol) and DBU (1 μ L, 0.006 mmol) in THF (0.1 mL) was stirred at room temperature for 2 h. The reaction mixture was poured into water and the product was extracted with AcOEt. The organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (7:1, v/v) as eluent to afford the title lactone 21 (5 mg, 90.9%) as a colorless gum: IR $\nu_{\rm max}$ 1700 cm⁻¹; ¹H NMR δ 0.75 (s, 3 H, 18-CH₃), 1.01 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.03 (s, 3 H, 19-CH₃), 1.88 and 1.94 (2 \times s, 2 \times 3 H, $27-CH_3$ and $28-CH_3$), 2.35-2.55 (m, 2 H, $23-CH_2$), 2.77 (t, J=3.1Hz, 1 H, 6-CH), 3.32 (s, 3 H, OCH₃), 4.38 (dt, J = 3.7 and 13.4 Hz, 1 H, 22-CH); HRMS calcd for $C_{29}H_{44}O_3$ (M⁺) 440.3291, found (M⁺) 440.3293. The spectroscopic data of 21 were identical with those reported.4c

(20S.22S.23S)-23-Acetoxy-3 β -hydroxywitha-5,24-dienolide (22). A mixture of 18 (50 mg, 0.1 mmol) and a catalytic amount of p-toluenesulfonic acid in aqueous dioxane (1 mL, dioxane: H₂O = 3:1) was heated at 80 °C for 1 h. The reaction mixture was poured into water, and the product was extracted with CH2Cl2. The organic layer was washed with aqueous saturated sodium bicarbonate solution and brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (2:1, v/v) as eluent afforded the alcohol 22 (48.5 mg, 99.8%) as a colorless amorphous: $[\alpha]^{21}_{\rm D}$ +129.1° (c 0.32, CHCl₃); mp 251–252 °C (hexane–CH₂Cl₂); IR $\nu_{\rm max}$ 3450, 1745, 1720 cm⁻¹; ¹H NMR δ 0.71 (s, 3 H, 18-CH₃), 1.01 (s, 3 H, 19-CH₃), 1.07 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.93 and 1.94 (2 × br s, 2 × 3 H, 27-CH₃ and 28-CH₃), 2.09 (s, 3 H, OAc), 3.45-3.6 (m, 1 H, 3-CH), 4.35 (dd, J = 2.4 and 5.5 Hz, 1 H, 22-CH), 5.34 (d, J = 5.5 Hz, 1 H, 6-CH), 5.46 (d, J= 2.4 Hz, 1 H, 23-CH); HRMS calcd for C₃₀H₄₄O₅ (M⁺) 484.3187, found (M⁺) 484.3192. Anal. Calcd for C₃₀H₄₄O₅: C, 74.34; H, 9.15. Found: C, 74.23; H, 9.36.

(20S,22S,23S)-23-Acetoxy-3 β -[(tert-butyldiphenylsilyl)oxy]witha-5,24-dienolide (23). A mixture of 22 (49 mg, 0.1 mmol), tert-butylchlorodiphenylsilane (53 µL, 0.2 mmol), triethylamine (35 µL, 0.25 mmol), and 4-(dimethylamino)pyridine (1.3 mg, 0.01 mmol) in CH₂Cl₂ (1 mL) was refluxed for 6 h. The reaction mixture was poured into water, and the product was extracted with CH2Cl2. The organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane–AcOEt (15:1, v/v) as eluent to afford 23 (72.5 mg, 99.2%) as a colorless gum: $[\alpha]^{23}_D$ +65.6° (c 0.38, CHCl₃); IR $\nu_{\rm max}$ 1730, 1700 cm⁻¹; ¹H NMR δ 0.68 (s, 3 H, 18-CH₃), 0.98 (s, 3 H, 19-CH₃), 1.04 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.06 (s, 9 H, t-Bu), 1.92 and $1.93 \ (2 \times \text{br s}, 2 \times 3 \text{ H}, 27\text{-CH}_3 \text{ and } 28\text{-CH}_3), 2.06 \ (\text{s}, 3 \text{ H}, OAc),$ 3.45-3.6 (m, 1 H, 3-CH), 4.33 (dd, J = 1.8 and 5.5 Hz, 1 H, 22-CH), 5.11 (d, J = 4.9 Hz, 1 H, 6-CH), 5.43 (d, J = 1.8 Hz, 1 H, 23-CH),7.3-7.45 (m, 6 H, Ph), 7.65-7.7 (m, 4 H, Ph). Anal. Calcd for C₄₆H₆₂O₅Si: C, 76.41; H, 8.64. Found: C, 76.40; H, 8.76.

(20S,22R,25ξ)-3β-[(tert-Butyldiphenylsilyl)oxy]witha-5,23-dienolide (24). Reductive deacetoxylation of 23 was performed using the procedure described for 19 to afford the olefin 24 (80.0%) as a colorless gum: IR $\nu_{\rm max}$ 1720 cm⁻¹; ¹H NMR δ 0.69 (s, 3 H, 18-CH₃), 0.83 and 0.84 (2 × d, 2 × J = 6.7 Hz, 3 H, 21-CH₃), 0.99 (s, 3 H, 19-CH₃), 1.06 (s, 9 H, t-Bu), 1.34 and 1.40 (2 × d, 2 × J = 7.3 Hz, 3 H, 27-CH₃), 1.77 (s, 3 H, 28-CH₃), 2.85-3.0 (m, 1 H, 25-CH), 3.45-3.6 (m, 1 H, 3-CH), 4.85-5.0 (m, 1 H, 22-CH), 5.12 (d, J = 4.9 Hz, 1 H, 6-CH), 5.4-5.5 (m, 1 H, 23-CH), 7.3-7.45 (m, 6 H, Ph), 7.65-7.7 (m, 4 H, Ph). Anal. Calcd for C₄₄H₆₀O₃Si: C, 79.47; H, 9.09. Found: C, 79.48; H, 9.28.

 $(20S,22R,25\xi)$ -3 β -Hydroxywitha-5,23-dienolide (25). A mixture of 24 (60 mg, 0.09 mmol) and 50% HF (100 μ L, 2.5 mmol) in MeCN (1 mL) was stirred for 4 h at room temperature. The reaction mixture was poured into water, and the product was extracted with CHCl3. The organic layer was washed with aqueous saturated sodium bicarbonate solution and brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (3:1, v/v) as eluent, affording 25 (70 mg, 93.3%) as a colorless gum: IR $\nu_{\rm max}$ 3350, 1720 cm⁻¹; ¹H NMR δ 0.72 (s, 3 H, 18-CH₃), 0.85 and $0.\overline{86}$ (2 × d, 2 × J = 6.7 Hz, 3 H, 21-CH₃), 1.01 (s, 3 H, 19-CH₃), 1.35 and 1.41 (2 × d, 2 × J = 7.3 Hz, 3 H, 27-CH₃), 1.79 (s, 3 H, 28-CH₃), 2.2-2.35 (m, 1 H, 7-CH), 2.85-3.0 (m, 1 H, 25-CH), 3.45-3.6 (m, 1 H, 3-CH), 4.85-5.0 (m, 1 H, 22-CH), 5.35 (d, J = 4.9 Hz, 1 H, 6-CH), 5.45-5.55 (m, 1 H, 23-CH); HRMScalcd for C₂₈H₄₂O₃ (M⁺) 426.3134, found (M⁺) 426.3139. Anal. Calcd for C₂₈H₄₂O₃: C, 78.82; H, 9.92. Found: C, 78.52; H, 10.09.

(20S,22R)-3 β -Hydroxywitha-5,24-dienolide (26). Isomerization of 25 was performed using the procedure described for 21 to afford the olefin 26 (98.4%) as a colorless plates: $[\alpha]^{24}_{\rm D}$ +39.2° (c 0.28, CHCl₃); mp 222–223 °C (hexane–CH₂Cl₂); IR $\nu_{\rm max}$ 3480, 1690 cm⁻¹; ¹H NMR δ 0.72 (s, 3 H, 18-CH₃), 1.01 (s, 3 H, 19-CH₃), 1.02 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.88 and 1.94 (2 × s, 2 × 3 H, 27-CH₃ and 28-CH₃), 3.45–3.6 (m, 1 H, 3-CH), 4.38 (dt, J = 3.7 and 13.4 Hz, 1 H, 22-CH), 5.35 (d, J = 4.9 Hz, 1 H, 6-CH); HRMS calcd for C₂₈H₄₂O₃ (M⁺) 426.3134, found (M⁺) 426.3137. Anal. Calcd for C₂₈H₄₂O₃: C, 78.82; H, 9.92. Found: C, 78.76; H, 10.09.

(20S,22R)-3-Oxowitha-5,24-dienolide (27). To a stirred solution of oxalyl chloride (15 µL, 0.17 mmol) in CH₂Cl₂ (0.2 mL) was added a solution of DMSO (14.5 μL, 0.21 mmol) in CH₂Cl₂ (0.2 mL) at -50 °C under argon atmosphere. After stirring for 15 min at the same temperature, the alcohol 26 (29 mg, 0.07 mmol) in CH₂Cl₂ (0.4 mL) was added and the reaction mixture was stirred for 30 min. Triethylamine (47 μ L, 0.34 mmol) was added, and stirring was further continued for 15 min at the same temperature. After addition of aqueous saturated NH₄Cl solution, the product was extracted with CH₂Cl₂. The organic layer was washed with brine and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane-AcOEt (4:1, v/v) as eluent to afford 27 (25.5 mg, 88.4%) as a colorless powder: $[\alpha]^{25}_{\rm D}$ +63.0° (c 0.18, CHCl₃); IR $\nu_{\rm max}$ 1690 cm⁻¹; ¹H NMR δ 0.75 (s, 3 H, 18-CH₃), 1.03 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.20 (s, 3 H, 19-CH₃), 1.89 and 1.94 (2 \times s, 2×3 H, 27-CH₃ and 28-CH₃), 2.83 (dd, J = 2.5 and 16.5 Hz, 1 H, 4-CHH), 3.2-3.35 (m, 1 H, 4-CHH), 4.38 (dt, J = 3.7 and 13.4Hz, 1 H, 22-CH), 5.3-5.4 (m, 1 H, 6-CH); HRMS calcd for C₂₈H₄₀O₃ (M⁺) 424.2978, found (M⁺) 424.2979.

Minabeolide-3 (28). A mixture of 27 (9 mg, 0.02 mmol) and oxalic acid (0.8 mg, 0.01 mmol) in EtOH (0.5 mL) was heated at 40 °C for 3 h. The reaction mixture was poured into water, and the product was extracted with AcOEt. The organic layer was washed with aqueous saturated sodium bicarbonate solution and brine, and dried over Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography on silica gel using hexane–AcOEt (3:1, v/v) as eluent to afford 28 (9 mg, 100%) as colorless needles: $(a)^{23}$ _D +131.6° (c 0.59, CHCl₃); mp 229–230 °C (CH₂Cl₂–AcOEt); IR $\nu_{\rm max}$ 1695, 1660 cm⁻¹; ¹H NMR δ 0.75 (s, 3 H, 18-CH₃), 1.01 (d, J = 6.7 Hz, 3 H, 21-CH₃), 1.19 (s, 3 H, 19-CH₃), 1.89 and 1.94 (2 × s, 2 × 3 H, 27-CH₃ and 28-CH₃), 2.2–2.6 (m, 4 H, 6-CH₂ and 23-CH₂), 4.37 (dt, J = 3.7 and 13.4 Hz, 1 H, 22-CH), 5.73 (br s, 1 H, 4-CH); HRMS calcd for $C_{28}H_{40}O_3$ (M⁺) 424.2978, found (M⁺) 424.2981. The ¹H NMR spectrum of 28 was identical with that reported.²

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